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## **Determining the Molar Volume of a Gas - A. Sedano - AP**

...

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**Molar Volume  
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Answers**

Molar volume =  
22.4141 L/mole at  
standard temperature  
(melting ice)  $T =$   
273.15 K and standard

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pressure  $p_0 = 1 \text{ ATM}$  ( $= 1.01325 \times 10^5 \text{ Pa}$ ) (At room temperature  $T = 298 \text{ K}$  and  $p = p_0$  the molar volume is  $24.5 \text{ L}$ ...

**LAB: THE MOLAR VOLUME OF A GAS**

moles, number of particles, molar volume at STP Chemistry Molar Volume at STP Practice Problems Assume all of the following problems are at standard conditions. 1. What

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volume will 0.750 moles of nitrogen gas occupy? 2. What volume will 100. grams of oxygen gas occupy? 3. How many moles are contained in 2.500 liters of methane ( $\text{CH}_4$ )? 4.

## **Molar Volume of a gas? | Yahoo Answers**

Chemistry. How do you calculate mass in grams from volume and molarity? Answer.



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Wiki User March 11, 2016 5:39PM. 1. You need to know the molar mass of the solute. ... (in kilograms per mole), and ...

**Which answer is NOT a measure of the amount of matter ...**

the molar volume is the volume occupied by one mole of any gas, at room temperature and pressure The molar volume is equal to 24

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dm<sup>3</sup> (24,000 cm<sup>3</sup>).

This volume is given in questions that need it....

## **Chemistry Molar Volume at STP Practice Problems**

The molar volume of hydrogen gas made is dependent of the number of moles of magnesium combined with excess hydrochloric acid. This is because "excess" means there is more

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hydrochloric acid than solid magnesium, and so the magnesium is the limiting reactant because it runs out first.

### **Molar Volume Quiz - MCQs Questions and Answers - College ...**

Molar volume is the volume occupied by one mole of a substance. This indicates that there is a 1:1 molar ratio of each gas. Asked in

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Chemistry, Atoms and  
Atomic Structure,  
Molecular Mass

**What is molar  
volume - Answers**

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(MCQs), molar volume,  
daltons law, gas

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properties, electron affinity, properties of crystalline solids quiz for SAT prep classes.

### **Molar gas volume - Higher - BBC Bitesize**

I need some help with this chem problem, don't know where to go next its a lab experiment to determine the molar volume of a gas and to verify that under STP its 22.4L materials

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used: 3.0M HCl

Magnesium strips flask  
that can hold 141mL  
gas (measured out)  
using pressure sensors  
and temperature  
sensors, after 3 tires  
the data i got was 1st  
try .08g Mg 5mL HCl  
the pressure went from  
100.7kPa to ...

## **Molar Volume: Using Avogadro's Law to Calculate the ...**

LAB: THE MOLAR  
VOLUME OF A GAS

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Answers

Introduction Today you are going to prove experimentally that the volume of one mole of a gas at standard temperature & pressure (STP) occupies a volume of 22.4 liters or 22,400 milliliters. The numerical values that are used for STP are one atmosphere (1 atm) and zero degrees ... Prelab Questions (answer key) 1 ...

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The molar volume of a gas expresses the



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volume occupied by 1 mole of that respective gas under certain temperature and pressure conditions. The most common example is the molar volume of a gas at STP (Standard Temperature and Pressure), which is equal to 22.4 L for 1 mole of any ideal gas at a temperature equal to 273.15 K and a pressure equal to 1.00 atm.

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**What is molar  
volume of a gas -  
Answers**

Molar volume =  
22.4141 L/mole at  
standard temperature  
(melting ice)  $T =$   
273.15 K and standard  
pressure  $p_0 = 1 \text{ ATM} (=$   
 $1.01325 \times 10^5 \text{ Pa})$  (At  
room temperature  
 $T = 298 \text{ K}$  and  $p = p_0$  the  
molar volume is 24.5  
L...

**Molar Volume of a  
Gas - Chemistry |**

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**Socratic**

Answer: STP is not the measure of amount of matter. Explanation: Matter is defined as the substance which takes place and has some mass.. For the given options: Molar mass is defined as the mass which is contained by 1 mole of a substance.It is considered as the unit of matter. Mole is defined as the mass of substance that has same number of

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fundamental units,  
which are present in  
12.000 g of ...

**How do you  
Calculate the molar  
volume of a gas -  
Answers**

The formula of molar  
volume is expressed as  
Where  $V_m$  is the  
volume of the  
substance. The  
standard temperature  
used is 273 Kelvin or 0  
o C, and standard  
pressure is 1

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atmospheres, i.e., 760 mm Hg.

Experimentally, one mole of any gas occupies a volume of 22.4 liters at STP.

**How do you calculate mass in grams from volume and ...**

Calculating volumes using masses, including using excess calculations and comparing of volumes directly. Useful

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playlist <https://www.y...>

## **Molar Volume Formula with Solved Examples**

The quiz is a series of questions about concepts related to molar volume. Some questions will deal with definitions. Other questions will require you to calculate the molar volume of a specific ...

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**What is molar  
volume at stp -  
Answers**

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room temperature  
 $T = 298 \text{ K}$  and  $p = p_0$  the  
molar volume is 24.5  
L...

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